

Ammonia as Chemical Energy Carrier: Electro-Catalytic Synthesis and Decomposition

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The talk focuses on alternative electro- and thermo-catalytic pathways for the production and decomposition of ammonia related to the use of NH₃ as carrier of renewable energy. Recent advances in electrocatalytic ammonia synthesis in proton-conducting ceramic cells (PCCs) are discussed with a focus on iron-based electrodes [1]. The effects of temperature, gas flow, voltage, and electrolyte thickness on electrochemical ammonia synthesis are investigated. To differentiate the various effects and mechanisms contributing to the electrocatalytic formation of NH₃, different gas flow configurations are studied. The experimental results demonstrate that NH₃ formation is primarily governed by the applied cell voltage, while the current density plays only a minor role. A strong interaction between electro- and thermo-catalytic reactions occurs. Co-feeding H₂ at the cathode proved advantageous for optimizing reaction conditions and increasing ammonia synthesis rates to values of $3 \times 10^{-8} \text{ mol s}^{-1} \text{ cm}^{-2}$ using a PCC with an iron based electrode of an active area of 12.57 cm². Also, decomposition of NH₃ to H₂ can be conducted in PCC. In all cases both electro- and thermos-catalytic processes have to be understood [2]. The combination of thermo-catalytic and electrochemically supported ammonia synthesis and decomposition opens new pathways for the electrification of NH₃ production using the rather inexpensive material iron and the use of NH₃ as hydrogen carrier.

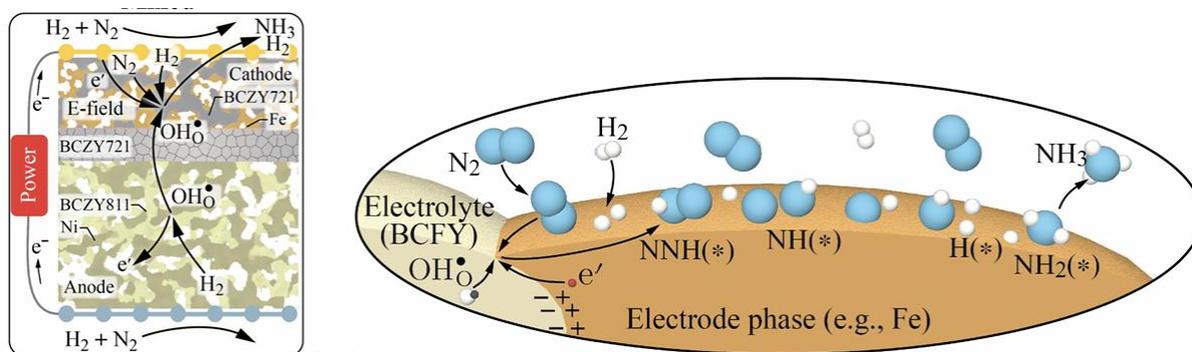


Figure 1: Schematic representation of reactions occurring at a proton-conducting ceramic cell that either contribute to NH₃ synthesis or compete with it. The reactions take place simultaneously; adapted from [1].

[1] P. Blanck, E.P. Martin, D. Schmider, J. Dailly, R.J. Kee, O. Deutschmann. Electrochemical Ammonia Synthesis in a Proton-conducting Ceramic Cell: A Parameter Study of an Iron-based Electrode. J. Electrochem. Soc. (2025) DOI: 10.1149/1945-7111/adfc9e.

[2] S. Davari, R. Chacko, T. Bastek, P. Lott, J. Dailly, S. Angeli, O. Deutschmann. Experimental and Microkinetic Investigation of Thermo-Catalytic Ammonia Decomposition over a Ba-promoted Ru/Ni BCZY Catalyst for Use in Ammonia-fed Protonic Ceramic Cells. Appl. Catal. A. 708 (2025) 120571. DOI:10.1016/j.apcata.2025.120571