

Effect of pre-treatment conditions on Fe-based catalyst for e-fuel production via modified Fischer-Tropsch synthesis

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The study investigates the effect of pre-treatment conditions on iron-based catalysts used for e-fuel production through a modified Fischer–Tropsch process [1-2]. The aim is to optimize the yield of liquid hydrocarbons from CO₂ by examining how different calcination and reduction atmospheres and temperatures affect catalyst stability and product distribution. The catalysts, consisting of Na-doped Fe₃O₄ [3], are tested in a fixed-bed reactor and characterized before and after catalytic runs. The results show that a higher promoter content enhances CO₂ conversion and selectivity toward long-chain hydrocarbons due to the formation of Hägg carbide (Fe₅C₂). However, excessive carburization accelerates catalyst deactivation. The optimal pre-treatment [4-5] therefore balances the formation of active phases to maximize the C₆–C₁₄ fraction, the olefin/paraffin ratio, and long-term catalyst stability (Figure 1). The integration of zeolites plays a key role by promoting hydrogenation and isomerization of light olefins, increasing the production of liquid fuels and supporting the economic feasibility of large-scale e-fuel production.

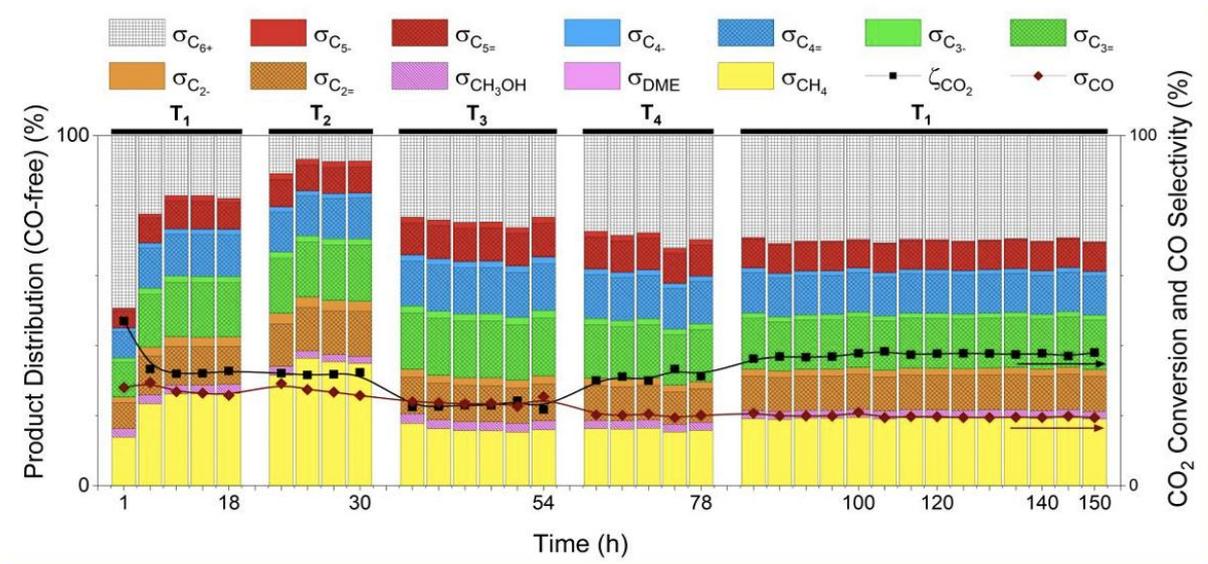


Figure 1 – Catalytic test after pre-treatment. GHSV 7 NL h⁻¹ gcat⁻¹, 300-400 °C, 23 bara, H₂/CO₂ = 3.

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